# **ChangeLog Preparatory Problems IChO 2023 V2**

# Problems (Theory):

- G0 General Instructions
  - Nernst equation was corrected (ln(Q) out of the fraction)
- Q5
  - Introduction: Oxidation state of RuO<sub>2</sub> was added explicitly in the Latimer Diagram
  - $\circ \quad 5.9 \text{ H2O} \rightarrow \text{H}_2\text{O}$
- Q7
  - 7.4 Added the hint that the deprotonation steps can be considered decoupled.
  - 7.7 <u>**Calculate</u>** how long it will take until the limiting value of 90  $\mu$ g/L is reached.  $\rightarrow$  <u>**Calculate**</u> how long it will take until the limiting value of 90  $\mu$ g/L free Pb is reached.</u>
  - Paragraph after 7.7 monitored over a period of 6 hours  $\rightarrow$  monitored by the concentration change in free Pb over a period of 6 hours
  - 7.8 Clarified the apparent half life and rate constant for the clearance of free Pd of the patients blood should be calculated.
  - 7.9 Clarified that the free Pb concentration is meant, and the complex can be ignored.
- Q19
  - $\circ$  Introduction: retention time  $\rightarrow$  reduced/adjusted retention time
- Q24
  - $\circ$   $\,$  Removed "Introductory Text" before introduction  $\,$
- Q30
  - Intro 30.1 The construction of the A/B ring fragment → The construction of the B/C ring fragment
  - $^\circ$  30.2 assembly of the A/B ring system in I  $\rightarrow$  assembly of the B/C ring system in I
  - 30.3 containing the C/D ring system both fragments were coupled → containing the A/D ring system both fragments were coupled:
  - Figure 35.5 Missing ester group in the product was added

#### Problems (Practical):

- Q6
  - Glassware and equipment: 25 ml round bottom flask  $\rightarrow$  50 ml round bottom flask
  - Glassware and equipment: Cork ring (for 25 mL round-bottom flask) → Cork ring (for 50 mL round-bottom flask)
  - Synthesis step 2.: <u>Clamp</u> a 25 mL round-bottom flask → <u>Clamp</u> a 50 mL round-bottom flask

## Solutions (Theory):

- Q6
  - $^\circ$   $\,$  6.5 Font color was changed and missing -1) was inserted in the last line
- Q7
  - $\circ~~$  7.4 Added justification for the approximation of decoupling the steps
  - $\circ~~$  7.5 Added missing methyl groups on O-methyl-DMSA
- Q9

- $\circ$  9.7 First checkbox was unchecked, the option was wrong as described in the text below
- Q23

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- 23.6 Structure **K** was corrected missing oxygen next to fluorenylmethyl was inserted Q25
- 25.2 Structure **A** was changed from a hemiacetal to a methyl acetal
- Q15,16 were designed better
- In general the solutions were designed a bit more consistently

#### **ChangeLog Preparatory Problems IChO 2023 V3**

#### Problems (Theory):

- Q2
  - $^\circ$  2.1 Inserted a 2 before  $H^{\scriptscriptstyle +}$  to be clear
- Q4
  - °  $4.6 \text{ S}^{\circ}(\text{ZnO}) = -43.6 \rightarrow \text{ S}^{\circ}(\text{ZnO}) = 43.6$
- 。 • Q5
  - Molar mass of Ru(H<sub>2</sub>O)<sub>3</sub>Cl<sub>3</sub> was corrected from 262.05 to 261.48 g/mol
  - ° Table 1 below 5.8: The last reaction equation was incorrectly balanced:  $MnO_4 + 2H_3O^+ + e^- \rightarrow H_2MnO_4 + \frac{3}{2}H_2O$
  - $^\circ$  Table 2 below 5.8: The charge in the last reaction equation was incorrect  $MnO^-4$  +  $e^- \rightarrow \ MnO^{32-}{}_4$
- Q18
  - 18.2 Stated pressure and temperature of oxygen stream (298 K and 100000 Pa)
  - Paragraph above 18.5 Specification that the particles are assumed to have a spherical shape
  - ° 18.5, 18.6 Corrected unit  $nm^{-3} \rightarrow nm^{3}$
- Q19
  - ° 19.2 XeO<sub>4</sub> → XeOF<sub>4</sub>
- Q29
  - 29.3 Vis under the reaction arrow was removed. This would be for the reverse reaction.

#### Solutions (Theory):

- Refinement to make them more consistent
- Q6
  - Natural abundance was interpreted wrong. The first line was replaced with the corrected mass